Mixtures

- · Keep their properties
- · Do not form new properties
- · Can be separated by:

hands sifting filters magnetism

Solutions

2 or more A special kind substances of mixture

Can be found everywhere

Both are mixtures Substances form

NEW physical properties

Can be separated by evaporation (saltwater)

Sedimentation

e.g. sand + water, dirt +

muddy water

When mixture is heterogeneous and solid constituent is heavy?

Homogenous Mixture and one constituent is heavy?

Centrifugation

e.g. separate cream from milk

Evaporation

e.g. common salt +

water, Sulphur in

Carbon disulphide

When solid present in the liquid is non-volatile?

Separating components of solid-liquid mixtures

When one the component dissolves in a solvent?

Filtration

e.g. common salt + sand

Any homogeneous or heterogenous mixture?

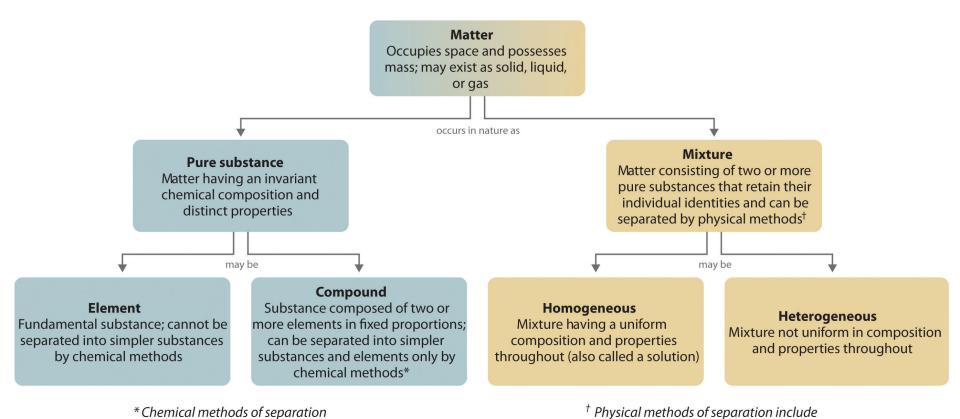
common salt + water

Distillation

When components are soluble in solvent?

Chromatography

Ink compoenents



filtration, distillation, and crystallization.

include electrolysis.